Nickel Scandium Germanide $Sc_3Ni_4Ge_4$: A New Ternary Germanide with the $Gd_6Cu_8Ge_8$ -Type Structure

BY B. YA. KOTUR* AND M. SIKIRICA

Laboratory of General and Inorganic Chemistry, Faculty of Science, University of Zagreb, PO Box 153, 41001 Zagreb, Yugoslavia

(Received 11 May 1981; accepted 25 September 1981)

Abstract. Sc₃Ni₄Ge₄, orthorhombic, *Immm*, Z = 2, a = 12.910 (9), b = 6.598 (4), c = 3.908 (2) Å, V = 332.88 Å³, $\mu = 31.1$ mm⁻¹ (Mo K \bar{a} , $\lambda = 0.7107$ Å). The compound crystallizes with the Gd₆Cu₈Ge₈-type structure. The structure was refined from single-crystal data. The final R = 0.042 for 205 reflections with isotropic thermal parameters for all atoms.

Introduction. During the investigation of the Sc-Ni-Ge system, among several new ternary germanides, a compound with the approximate composition $Sc_3Ni_4Ge_4$ was found. Its powder diffractogram was very similar to that of $Sc_3Ni_4Si_4$ (Bodak, Kotur & Gladyshevsky, 1976), the crystal structure of which was reported to be of the Gd₆Cu₈Ge₈ type (Rieger, 1970). As the crystal structure of $Sc_3Ni_4Si_4$ had been determined from powder data and the atomic parameters had not been refined, we decided to perform a full structure investigation of $Sc_3Ni_4Ge_4$ from single-crystal data.

A single crystal of prismatic shape with dimensions $0.024 \times 0.008 \times 0.066$ mm was isolated from a crushed ingot of stoichiometry close to Sc₃Ni₄Ge₄. The sample was prepared by arc melting. The single crystal was examined by Laue and rotation photographs. The Laue class *mmm* and the systematic absences (*hkl* with h + k + l = 2n, *hkO* with h + k = 2n, *hOl* with h + l = 2n, 0kl with k + l = 2n, *hOO* with h = 2n, 0kO with k = 2n, 00l with l = 2n) indicate *Immm*, *Imm2*, *I222* and $I2_12_12_1$ as the possible space groups. The space-group symmetry together with the lattice parameters indicated the Gd₆Cu₈Ge₈-type structure.

Intensity data were collected by means of a Philips PW 1100 four-circle diffractometer (graphite-monochromatized Mo Ka radiation). Integrated intensities of 205 reflections were collected in the θ -2 θ scan mode up to a limit of sin $\theta/\lambda = 0.70$ Å⁻¹. Lattice parameters (see *Abstract*) were obtained from the 2 θ values for 13 reflections by means of the least-squares method.

The parameters of $Gd_6Cu_8Ge_8$ given by Rieger (1970) were used as starting values for a least-squares

refinement with the program SYST (Domenicano, Spagna & Vaciago, 1969). Corrections for Lorentz– polarization effects and for absorption were made. The atomic scattering factors were those of Cromer & Mann (1968) with corrections for anomalous scattering from Cromer & Liberman (1970). The final $R = \sum |\Delta F| / \sum F_o$ was 0.042.*

The final atomic parameters are given in Table 1 and interatomic distances in Table 2.

* Lists of structure factors have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 36436 (3 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Positional $(\times 10^4)$ and isotropic thermal parameters for Sc₃Ni₄Ge₄

E.s.d.'s are given in parentheses.

		x	у	z	$B_{\rm iso}$ (Å ²)
Sc(1)	2(d)	0	$\frac{1}{2}$	0	0.564 (6)
Sc(2)	4(e)	1266 (3)	Õ	0	0.35 (4)
Ni	8(n)	3231 (2)	1960 (3)	0	0.39 (3)
Ge(1)	4(f)	2189 (2)	1	0	0.373(5)
Ge(2)	4(h)	0	194Ĩ (4)	$\frac{1}{2}$	0.32 (4)

Table 2. Interatomic distances (Å) in $Sc_3Ni_4Ge_4$

E.s.d.'s are given in parentheses.

Sc(1)-4Sc(2)	3.682 (3)	Ni-2Sc(1)	3.272 (2)
8Ni	3.272 (2)	2Sc(2)	2.875 (4)
2Ge(1)	2.826(3)	Sc(2)	2.847 (5)
4Ge(2)	2.809 (2)	2Ni	2.809 (3)
		Ni	2.586 (3)
Sc(2)-2Sc(1)	3.682 (3)	Ge(1)	2.415 (3)
2Ge(1)	3.508 (3)	2Ge(1)	2.405 (2)
Sc(2)	3.269 (6)	Ge(2)	2.396 (3)
4Ni	2.875 (4)		
4Ge(2)	2.851 (3)	Ge(1)-2Sc(2)	3.508 (3)
2Ni	2.847 (5)	Sc(1)	2.826 (3)
2Ge(1)	2.792 (4)	2Sc(2)	2.792 (4)
		2Ni	2.415 (3)
		4Ni	2.405 (2)
		Ge(2)-4Sc(2)	2.851 (3)
		2Sc(1)	2.809 (2)
		Ge(2)	2.561 (4)
		2Ni	2.396 (3)

© 1982 International Union of Crystallography

^{*} Permanent address: Chair of Inorganic Chemistry, University of Lvov, Lomonosov Street 6, 290005 Lvov, USSR.

Discussion. $Sc_3Ni_4Ge_4$ is a new representative of a family of Gd₆Cu₈Ge₈-type structure phases. The compounds with this structure were found previously in the ternary systems R-Cu-Si(Ge) (R = rare-earthmetal, REM) (Rieger, 1970; Hanel & Nowotny, 1970) and Zr(Hf)-Cu-Si (Springer, 1974). The radius of the largest atom in these compounds varies in the range 1.80 (Sm) to 1.59 (Hf) Å. The same is true for the ThCu₂Si₂ (Ban & Sikirica, 1965) and TiNiSi (Shoemaker & Shoemaker, 1965) structure types. This is why families of compounds with these three types of structure are numerous. The phases with the ThCu₂Si₂ and TiNiSi structure types are very often in equilibrium with the phase with Gd₆Cu₈Ge₈ type; they are its nearest neighbours. The compositions of these phases in the ternary systems R-T-T' (R = REM, Zr, Hf; T= 3d transition metal; T' = Si, Ge) lie on the section with a T-to-T' atomic ratio of 1:1. At least one of the lattice parameters of these structures is ~ 4 Å, which is the height of the trigonal prism which forms the coordination polyhedron around the T' atoms. Three kinds of trigonal prisms can be found in these structures: $[T'R_6]$, $[T'R_4T_2]$ and $[T'R_2T_4]$. The connections of the prisms in these structures are presented in Fig. 1. In some REM-containing systems



Fig. 1. Connections of trigonal prisms $[T'R_6]$, $[T'R_4T_2]$ and $[T'R_2T_4]$ in the structures isotypic with ThCu₂Si₂, Gd₆Cu₈Ge₈, TiNiSi and AlB₂.

(except Sc) the $Gd_6Cu_8Ge_8$ -type phases are in equilibrium with the phases of the AlB₂ type (Rieger, 1970). This is not surprising, since the former also contain fragments of the latter (see Fig. 1).

Thus on the section with atomic ratio T:T' = 1:1, the change of the R-atom content leads to a change in the crystal structure of the phases. The known factors which govern the occurrence and crystal structure of intermetallics are atomic size, electron concentration, electronegativity, etc. (Pearson, 1972). However, there are no quantitive methods, based on the factors mentioned above, which allow the theoretical calculation of the compositions and crystal structures of intermetallics. But it can be seen that the phases which lie on the same sections, defined by the atomic ratios R:T, R:T', T:T' etc., have interconnected structures (for one another, e.g. of the Sc: Si = 1:1 section in the Sc-Co-Si system, see Gladyshevsky & Kotur, 1978). In this way the search for geometric relationships among the structures may facilitate the investigation of the unknown structures.

The authors thank Mr R. I. Andrusyak for assistance in the preparation of the samples and Mr M. Bruvo for the diffractometer-data collection.

References

- BAN, Z. & SIKIRICA, M. (1965). Acta Cryst. 18, 594-599.
- BODAK, O. I., KOTUR, B. YA. & GLADYSHEVSKY, E. I. (1976). Dopov. Akad. Nauk Ukr. RSR Ser. A, pp. 655-657.
- CROMER, D. T. & LIBERMAN, D. (1970). J. Chem. Phys. 53, 1891–1898.
- CROMER, D. T. & MANN, J. B. (1968). Acta Cryst. A24, 321–324.
- DOMENICANO, A., SPAGNA, R. & VACIAGO, A. (1969). Atti Accad. Naz. Lincei Cl. Sci. Fis. Mat. Nat. Sez. 2a, 47, 331–336.
- GLADYSHEVSKY, E. I. & KOTUR, B. YA. (1978). Kristallografiya, 23, 946–950.
- HANEL, G. & NOWOTNY, H. (1970). Monatsh. Chem. 101, 463–468.
- PEARSON, W. B. (1972). The Crystal Chemistry and Physics of Metals and Alloys. New York: John Wiley.
- RIEGER, W. (1970). Monatsh. Chem. 101, 449-462.
- SHOEMAKER, C. B. & SHOEMAKER, D. P. (1965). Acta Cryst. 18, 900–905.
- Springer, H. (1974). J. Less-Common Met. 34, 39-71.